

# Vibrational structure of absorption spectrum of 2,5-bis(2-benzoxazolyl)hydroquinone

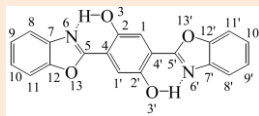
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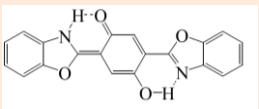
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## Introduction

Photoinduced excited state intramolecular proton transfer (ESIPT) has attracted attention of researchers for several decades as an ultrafast photochemical reaction and a process that could be utilized for a number of applications. 2,5-bis(2-benzoxazolyl)hydroquinone (BBHQ) has been extensively studied for several decades by theoretical calculations and experiments in various environments as a substance that exhibits the ESIPT. The preferable structure of the BBHQ molecule in the ground state is an enol form, whereas the enol and keto forms are in dynamical equilibrium in the first excited state, the equilibrium is shifted towards to the keto form. In this study we consider application of TDDFT approach and harmonic approximation for vibronic states of the enol form of the molecule of BBHQ.



enol form



keto form

## Computational methods

The calculations were performed for an isolated molecule using the ORCA software package [1,2], the def2-SVP basis set and  $\omega$ B97X-D3 functional. The vibrations were calculated in harmonic approximation using analytical second energy derivatives for the ground state and numerical ones with default parameters for the excited states. Frequencies of the vibrations are unscaled. Vibronic structure of the absorption spectra was calculated with the harmonic model and the Franck-Condon approximation. The vibronic structure calculations were performed with the FCClasses3 code [3,4] using the time-independent and time-dependent methods. The spectrum was modeled assuming inhomogeneous broadening by Gaussian function with the half width at half maximum of 0.035 eV.

## Conclusions

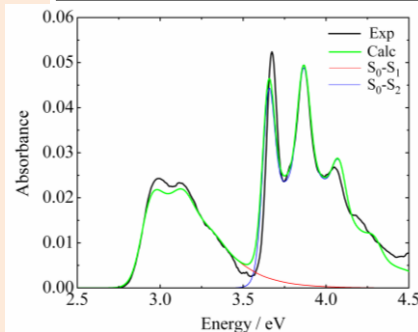
The calculations performed with the  $\omega$ B97X-D3 density functional for molecular vibrations and electronic transitions are found to reproduce well the vibronic structure of the  $S_0-S_1$  and  $S_0-S_2$  absorption bands in the spectrum of 2,5-bis(2-benzoxazolyl)hydroquinone measured in solution at room temperature. It allows to determine normal modes that contribute significantly to the absorption spectrum. At the same time energy of the electronic transitions is overestimated by the calculations and the overestimation is different for the  $S_0-S_1$  and  $S_0-S_2$  transitions.

## References

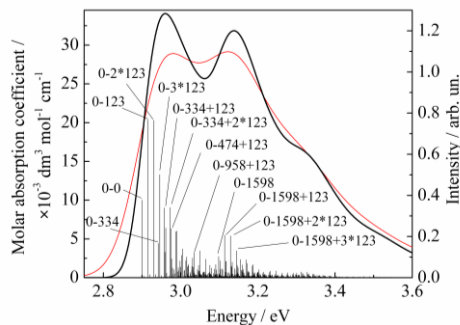
1. F. Neese, WIREs Comput. Mol. Sci., 2012, 2, 73.
2. F. Neese, WIREs Comput. Mol. Sci., 2017, 8, e1327.
3. F. Santoro, J. Cerezo, FCClasses3, A code for vibronic calculations. Version 3.0.3. 2023. <http://www.iccom.cnr.it/en/fcclasses>.
4. J. Cerezo, F. Santoro, J. Comput. Chem., 2023, 44, 626, doi:10.1002/jcc.27027.

Calculated geometrical parameters (bond length, Å; angle, degree) of the enol structure of the molecule of BBHQ in the ground  $S_0$ , first  $S_1$  and second  $S_2$  excited states

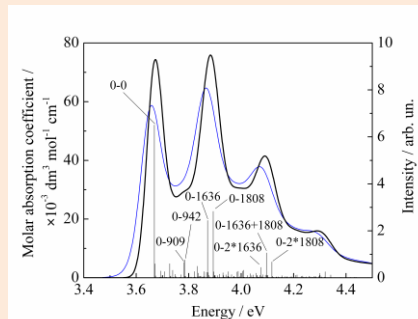
Parameter	$S_0$	$S_1$	$S_2$	Parameter	$S_0$	$S_1$	$S_2$	Parameter	$S_0$	$S_1$	$S_2$
$C_1-C_2$	1.393	1.403	1.386	$C_{12}-C_7$	1.398	1.404	1.412	$O_{13}-C_5-N_6$	114.6	114.0	113.7
$C_2-C_4$	<b>1.418</b>	<b>1.457</b>	1.425	$C_{12}-O_{13}$	1.363	1.365	1.357	$C_7-C_8-C_9$	117.1	117.3	117.3
$C_4-C_{11}$	<b>1.399</b>	1.401	<b>1.443</b>	$O_{13}-C_5$	1.352	1.358	1.367	$C_8-C_9-C_{10}$	121.7	121.8	121.9
$C_4-C_5$	<b>1.457</b>	<b>1.422</b>	<b>1.409</b>	$C_2-O_3$	<b>1.336</b>	<b>1.310</b>	1.335	$C_9-C_{10}-C_{11}$	121.7	121.5	121.6
$C_5-N_6$	<b>1.300</b>	<b>1.321</b>	<b>1.329</b>	$C_1-C_2-C_4$	<b>117.9</b>	<b>120.5</b>	118.4	$C_{10}-C_{11}-C_{12}$	115.6	115.9	115.8
$N_6-C_7$	<b>1.389</b>	1.377	<b>1.365</b>	$C_2-C_4-C_5$	<b>118.7</b>	<b>118.2</b>	<b>119.8</b>	$C_{11}-C_{12}-C_7$	123.8	123.7	123.8
$C_7-C_8$	1.397	1.398	1.405	$C_4-C_5-N_6$	126.0	125.3	126.1	$C_{12}-C_7-C_8$	120.1	119.8	119.6
$C_8-C_9$	1.391	1.393	1.389	$C_5-N_6-C_7$	105.1	105.3	105.2	$C_4-C_2-O_3$	<b>123.4</b>	<b>120.7</b>	122.5
$C_9-C_{10}$	1.407	1.405	1.41	$N_6-C_7-C_8$	107.8	107.9	108.4	$C_2-C_4-C_{11}$	<b>120.7</b>	<b>118.4</b>	<b>119.1</b>
$C_{10}-C_{11}$	1.393	1.397	1.397	$C_7-C_{12}-O_{13}$	107.5	107.8	107.7	$C_4-C_1-C_2$	<b>121.4</b>	121.1	<b>122.6</b>
$C_{11}-C_{12}$	1.388	1.384	1.384	$C_{12}-O_{13}-C_5$	105.0	104.9	105.0	-	-	-	-



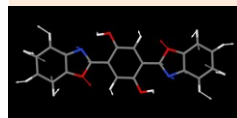
Absorption spectrum of BBHQ dissolved in  $CCl_4$  measured at room temperature (Exp) and spectrum modeled with the time-dependent method (Calc). Calculated energy of the  $S_0-S_1$  transition is red-shifted by 0.2 eV,  $S_0-S_2$  transition is red-shifted by 0.455 eV.



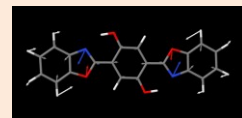
Absorption spectrum for the  $S_0-S_1$  transitions modeled with the time-dependent method at room temperature (red line) and time-independent method at temperature 0 K (black line). The stick spectrum represents vibronic transitions at 0 K. The spectra are red-shifted by 0.2 eV in energy.



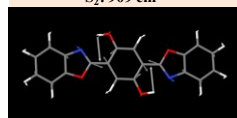
Absorption spectrum for the  $S_0-S_2$  transitions modeled with the time-dependent method at room temperature (blue line) and time-independent method at temperature 0 K (black line). The stick spectrum represents vibronic transitions at 0 K. The spectra are red-shifted by 0.455 eV in energy.



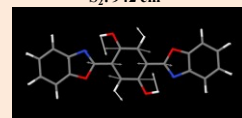
$S_2$ : 909  $cm^{-1}$



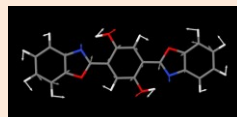
$S_2$ : 942  $cm^{-1}$



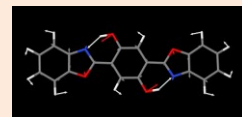
$S_2$ : 1636  $cm^{-1}$



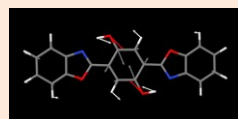
$S_2$ : 1808  $cm^{-1}$



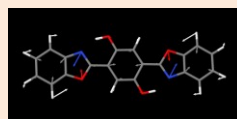
$S_1$ : 123  $cm^{-1}$



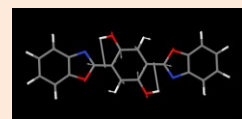
$S_1$ : 334  $cm^{-1}$



$S_1$ : 474  $cm^{-1}$



$S_1$ : 958  $cm^{-1}$



$S_1$ : 1598  $cm^{-1}$